

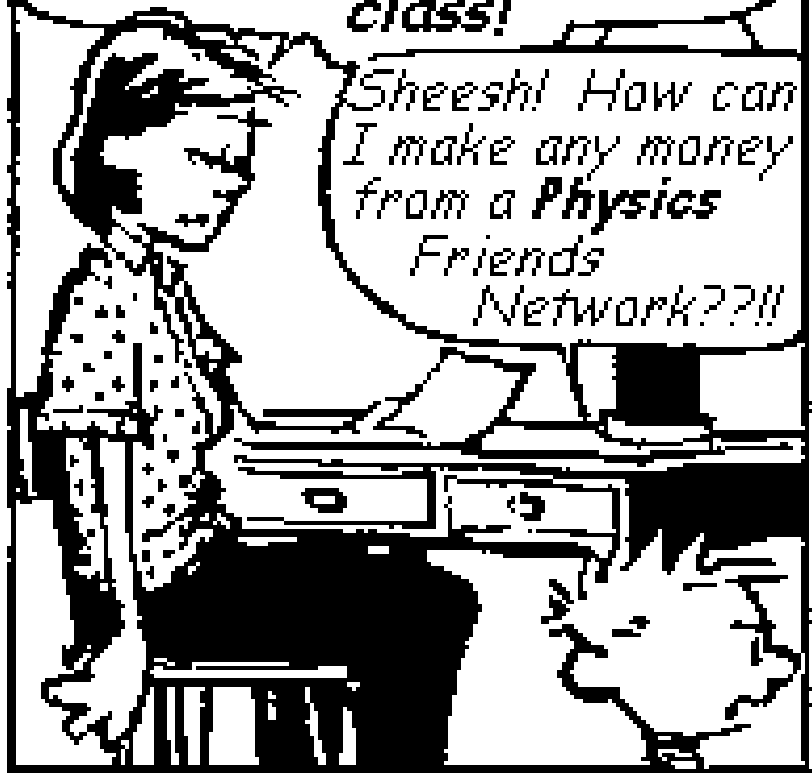
Ch. 1, Physics & Measurement

Guess what? I signed up for a psychics class at school. I'll be predicting stuff in no time!



*No, Calvin. You signed up for **physics** - a very challenging science class!*

*Sheesh! How can I make any money from a **Physics Friends Network**??!!*



Outline

Ch. 1, Physics & Measurement

1. Physics is an experimental science \Rightarrow Measurements

\Rightarrow **Units**

2. Physics is a quantitative science \Rightarrow Mathematics

\Rightarrow **Algebra & Calculus**

3. International System (SI) of units: Length \Rightarrow **m**,

Time \Rightarrow **s**, Mass \Rightarrow **kg**

Objects have different sizes \Rightarrow **Conversion of units**

4. Other properties of matter require the use of derived units:

$\rho = m/V$ \Rightarrow density unit: kg/m^3

If you know the formula \Rightarrow You can find the units

If you know the units \Rightarrow You can “guess” the formula!

Physics:

The most basic of all sciences!

- **Physics:** The “Mother” of all sciences!
- **Physics** = The study of the behavior of and the structure of matter and energy and of the interaction between matter and energy.

Physics: General Discussion

- **Goal of Physics** (& all of science): To quantitatively and qualitatively **describe** the “**world around us**”.
- **Physics IS NOT** merely a collection of facts and formulas!
- **Physics IS** a creative activity!
- Physics → Observation → Explanation.
- Requires **IMAGINATION!!**

Physics & its relation to other fields

- The “Mother” of all Sciences!
- The foundation for and is connected to **ALL** branches of *science and engineering*.
- Also useful in everyday life and in **MANY** professions
 - Chemistry
 - Life Sciences
 - Architecture
 - Engineering.
 -

Theory

- *Quantitative description* of experimental observations.
- Not just WHAT is observed but WHY it is observed as it is and HOW it works the way it does.
- Tests of theories:
 - Experimental observation:
More experiments, more observation.
 - Predictions:
Made before observations & experiments.

Model, Theory, Law

- **Model:** An analogy of a physical phenomenon to something we are familiar with.
- **Theory:** More detailed than a model. Puts the model into mathematical language.
- **Law:** Concise & **general** statement about **how nature behaves**. Must be verified by many, many experiments! Only a few laws.
 - Not comparable to laws of government!

Measurement & Uncertainty

- Physics is an *EXPERIMENTAL* science!
 - Finds relations between physical quantities.
 - Expresses those relations in the language of mathematics. (*LAWS & THEORIES*)
- Experiments are *NEVER* 100% accurate.
 - *Always* have *uncertainty* in final result.
 - Experimental error.
 - Common to state this precision (when known).

- Consider a simple measurement of the width of a board. Find **23.2 cm**.
- However, measurement is only accurate to 0.1 cm (estimated).

⇒ Write width as **(23.2 ± 0.1) cm**

± 0.1 cm ≡ *Experimental uncertainty*

- **Percent Uncertainty:**

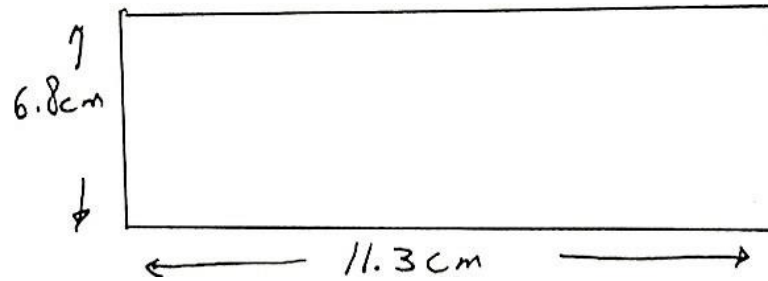
$$\pm (0.1/23.2) \times 100 \approx \pm 0.4\%$$

Significant Figures

- **Significant figures** (“sig figs”) \equiv The number of reliably known digits in a number.
- Calculation involving several numbers.
 - Number of sig figs in result \equiv Number of sig figs of the number containing the smallest number of sig figs which entered the calculation.

- Example:

(Not to scale!)



- Area of board, dimensions 11.3 cm \times 6.8 cm

- Area = (11.3) \times (6.8) = 76.84 cm²

11.3 has 3 sig figs , 6.8 has 2 sig figs

\Rightarrow 76.84 has too many sig figs!

- Proper number of sig figs in answer = 2

\Rightarrow Round off 76.84 & keep only 2 sig figs

\Rightarrow **Reliable answer for area = 77 cm²**

Sig Figs

- **General Rule:** Final result of multiplication or division should have only as many sig figs as the number with least sig figs in the calculation.
- **NOTE!!!!** All digits on your calculator are *NOT* significant!!

Powers of 10

(Scientific Notation)

- **READ Appendix B.1**
- Common to express very large or very small numbers using powers of 10 notation.
- Examples:

$$39,600 = 3.96 \times 10^4$$

(moved decimal 4 places to left)

$$0.0021 = 2.1 \times 10^{-3}$$

(moved decimal 3 places to right)

Units, Standards, SI System

- All measured physical quantities have units.
- Units are **VITAL** in physics!!
- In this course (and in most of the **modern world, *except the USA!***) we will use (almost) exclusively the **SI system of units**.

SI = “**Système International**” (French)

More commonly called the “**MKS system**” (meter-kilogram-second) or more simply, “**the metric system**”

SI or MKS System

- Defined in terms of **standards** for length, mass (we'll discuss later), and time.
- **Length unit: Meter (m)** (kilometer = km = 1000 m)
 - **Standard meter**. Newest definition in terms of speed of light \equiv Length of path traveled by light in vacuum in $(1/299,792,458)$ of a second!
- **Time unit: Second (s)**
 - **Standard second**. Newest definition \equiv time required for 9,192,631,770 oscillations of radiation emitted by cesium atoms!
- **Mass unit: Kilogram (kg)**
 - Discussed in detail later

Larger & smaller units defined from SI standards by powers of 10 & Greek prefixes

TABLE 1-4
Metric (SI) Prefixes

Prefix	Abbreviation	Value
exa	E	10^{18}
peta	P	10^{15}
tera	T	10^{12}
giga	G	10^9
mega	M	10^6
kilo	k	10^3
hecto	h	10^2
deka	da	10^1
deci	d	10^{-1}
centi	c	10^{-2}
milli	m	10^{-3}
micro [†]	μ	10^{-6}
nano	n	10^{-9}
pico	p	10^{-12}
femto	f	10^{-15}
atto	a	10^{-18}

[†] μ is the Greek letter “mu.”

Typical Lengths (approx.)

TABLE 1-1

Some typical Lengths or Distances (order of magnitude)

Length (or distance)	Meters (approximate)
Neutron or proton (radius)	10^{-15} m
Atom	10^{-10} m
Virus [see Fig. 1-6]	10^{-7} m
Sheet of paper (thickness)	10^{-4} m
Finger width	10^{-2} m
Football field length	10^2 m
Mt. Everest height [see Fig. 1-6]	10^4 m
Earth diameter	10^7 m
Earth to Sun	10^{11} m
Nearest star, distance	10^{16} m
Nearest galaxy	10^{22} m
Farthest galaxy visible	10^{26} m

Typical Times & Masses (approx.)

TABLE 1-2 Some typical Time Intervals

Time interval	Seconds (approximate)
Lifetime of very unstable particle	10^{-23} s
Lifetime of radioactive elements	10^{-22} s to 10^{28} s
Lifetime of muon	10^{-6} s
Time between human heartbeats	10^0 s (= 1 s)
One day	10^5 s
One year	3×10^7 s
Human life span	2×10^9 s
Length of recorded history	10^{11} s
Humans on Earth	10^{14} s
Life on Earth	10^{17} s
Age of Universe	10^{18} s

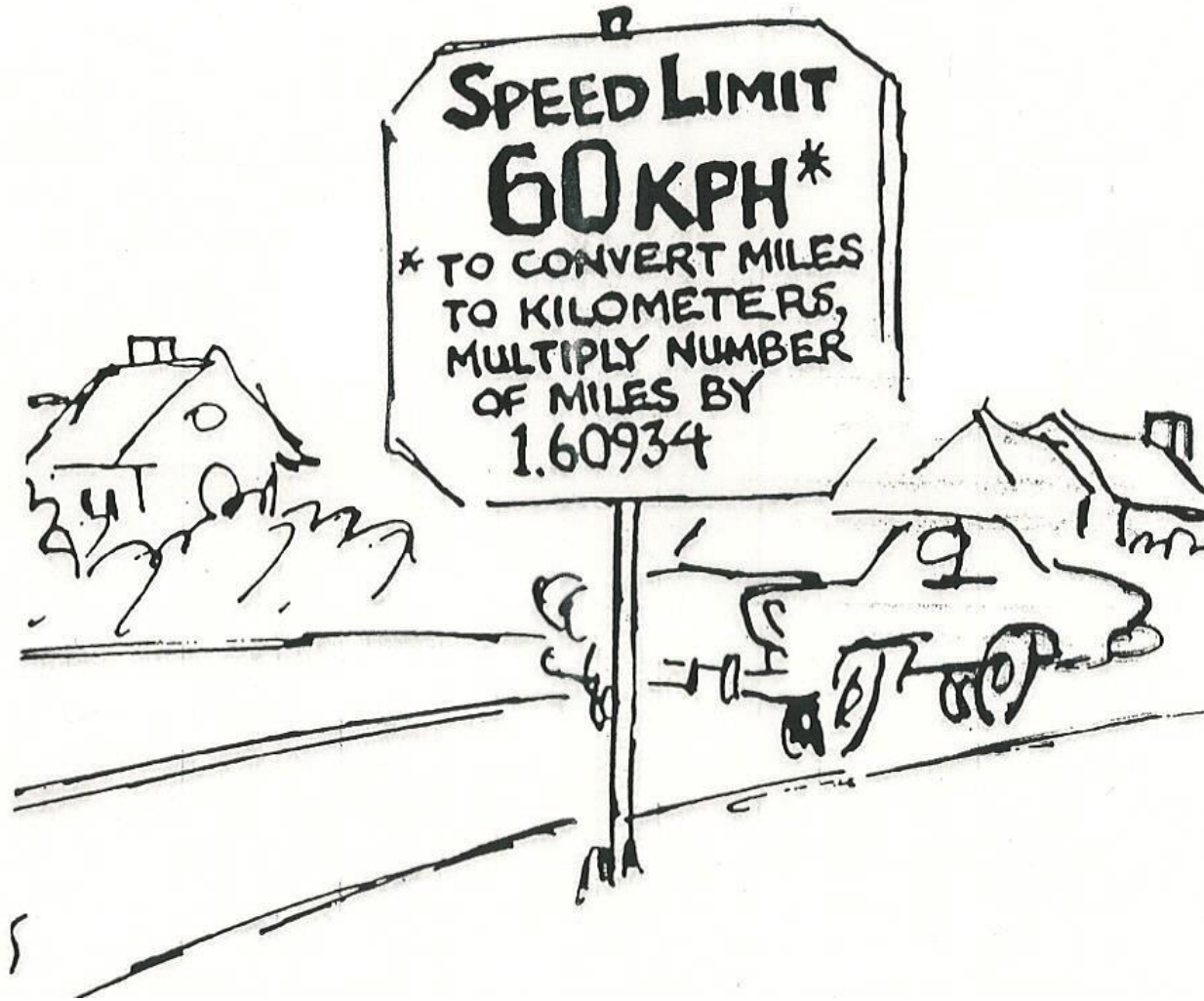
TABLE 1-3 Some Masses

Object	Kilograms (approx.)
Electron	10^{-30} kg
Proton, neutron	10^{-27} kg
DNA molecule	10^{-17} kg
Bacterium	10^{-15} kg
Mosquito	10^{-5} kg
Plum	10^{-1} kg
Person	10^2 kg
Ship	10^8 kg
Earth	6×10^{24} kg
Sun	2×10^{30} kg
Galaxy	10^{41} kg

Other Systems of Units

- **CGS** (centimeter-gram-second) system
 - **Centimeter** = 0.01 meter
 - **Gram** = 0.001 kilogram
- **British** (foot-pound-second) system
 - Our “everyday life” system of units
 - Only used by USA and some third world countries. Rest of world (including Britain!) uses SI system. We *will not* use the British System in this course!
 - Conversions exist between the British & SI systems. We *will not* use them in this course!

In this class, we will **NOT** do unit conversions!



We will work **exclusively** in SI (MKS) units!

Basic & Derived Quantities

- **Basic Quantity** \equiv Must be defined in terms of a standard (meter, kilogram, second).
- **Derived Quantity** \equiv Defined in terms of combinations of basic quantities
 - Unit of speed = meter/second = m/s

Units and Equations

- In dealing with equations, remember that the *units must be the same on both sides of an equation* (otherwise, *it is not an equation*)!
- **Example:** You go 90 km/hr for 40 minutes.
How far did you go?
 - Equation from Ch. 2: $x = vt$, $v = 90 \text{ km/hr}$, $t = 40 \text{ min}$. To use this equation, first convert t to hours:
 $t = (\frac{2}{3})\text{hr}$ so, $x = (90 \text{ km/hr}) \times [(\frac{2}{3})\text{hr}] = 60 \text{ km}$
The hour unit (**hr**) has (literally) cancelled out in the numerator & denominator!

Changing Units

- As in the example, units in the numerator & the denominator can cancel out (as in algebra)

- **Illustration:** Convert 80 km/hr to m/s

Conversions: 1 km = 1000 m; 1hr = 3600 s

⇒ 80 km/hr =

$$(80 \text{ km/hr}) (1000 \text{ m/km}) (1\text{hr}/3600 \text{ s})$$

(Cancel units!)

$$\mathbf{80 \text{ km/hr} \cong \mathbf{22 \text{ m/s}}}$$
 (22.222...m/s)

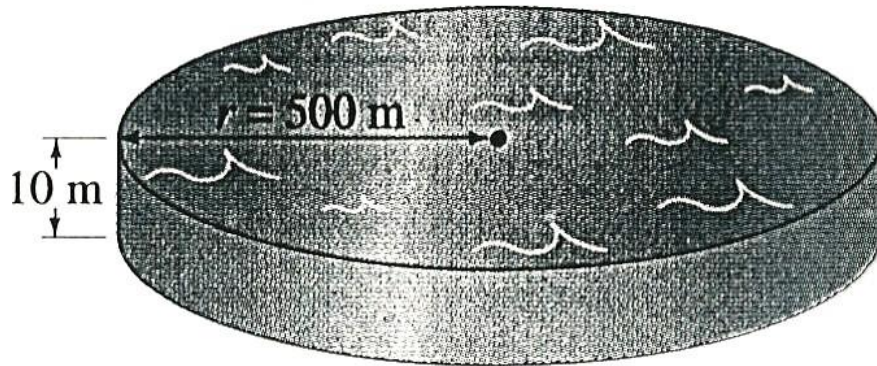
- Useful conversions:

$$1 \text{ m/s} \cong 3.6 \text{ km/hr}; 1 \text{ km/hr} \cong (1/3.6) \text{ m/s}$$

Order of Magnitude; Rapid Estimating

- Sometimes, we are interested in only an approximate value for a quantity. We are interested in obtaining rough or **order of magnitude estimates**.
- **Order of magnitude estimates:** Made by rounding off all numbers in a calculation to 1 sig fig, along with power of 10.
 - Can be accurate to within a factor of 10 (often better)

Example: $V = \pi r^2 d$



(b)

(a) How much water is in this lake?
(Photo is of one of the Rae Lakes in the Sierra Nevada of California.)

(b) Model of the lake as a cylinder.

[We could go one step further and estimate the mass or weight of this lake. We will see later that water has a density of 1000 kg/m^3 , so this lake has a mass of about $(10^3 \text{ kg/m}^3)(10^7 \text{ m}^3) \approx 10^{10} \text{ kg}$, which is about 10 billion kg or 10 million metric tons. (A metric ton is 1000 kg, about 2200 lbs, slightly larger than a British ton, 2000 lbs.)]

- Example: Estimate!